

Automatic Differentiation and Source Code Generation for Dynamic Modeling and Simulation of Brine Systems

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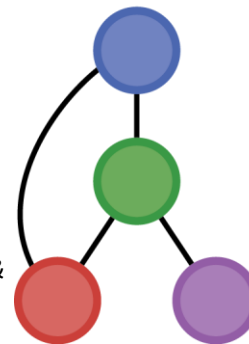
PI : Prof. Matthew Stuber (UConn)

Session : Applied Math for Energy and Environmental Applications

Nov 9th, 2023

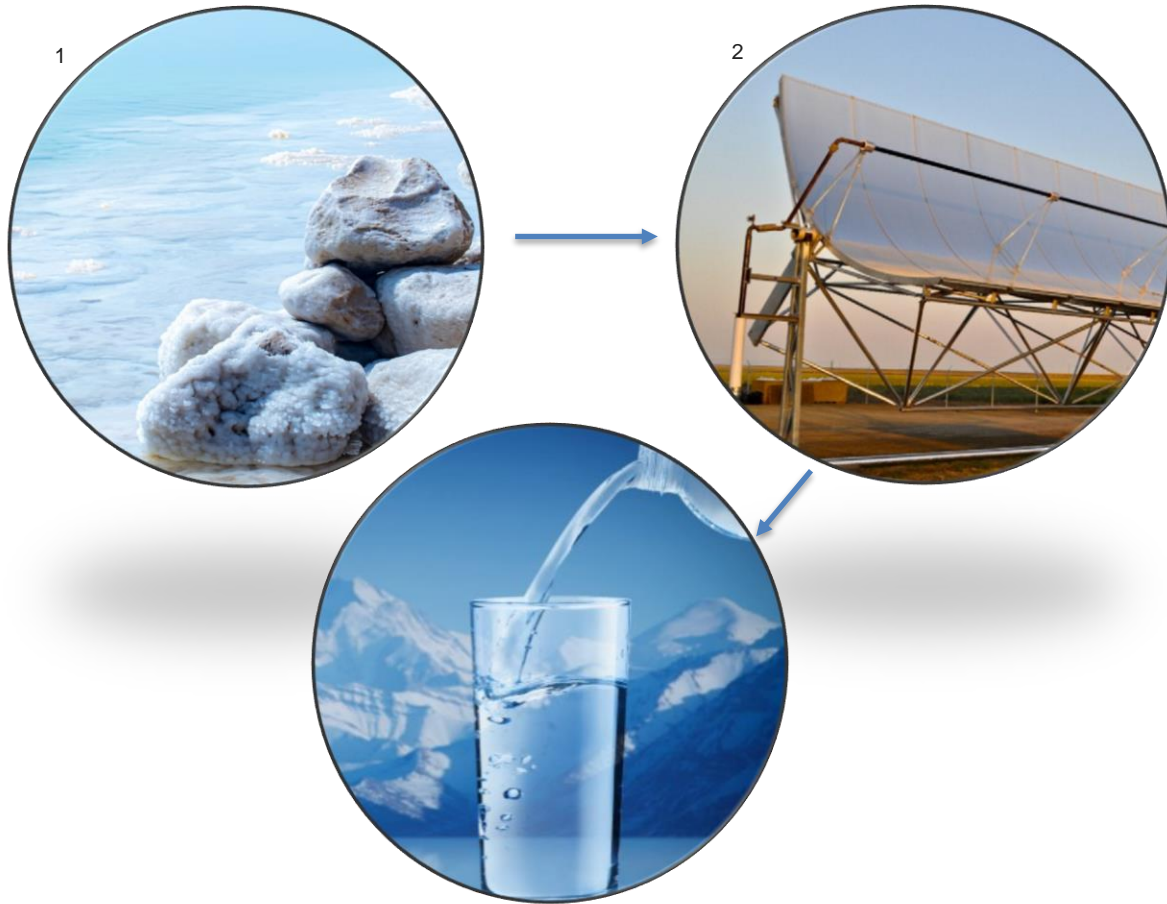
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Process Systems and
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Motivation



Thermal brine separation is of critical importance to many industries with brine effluent streams and/or brine concentration needs (e.g., agriculture, power production, mining)

- Increase sustainability
- Reduce costs
- Improve system robustness

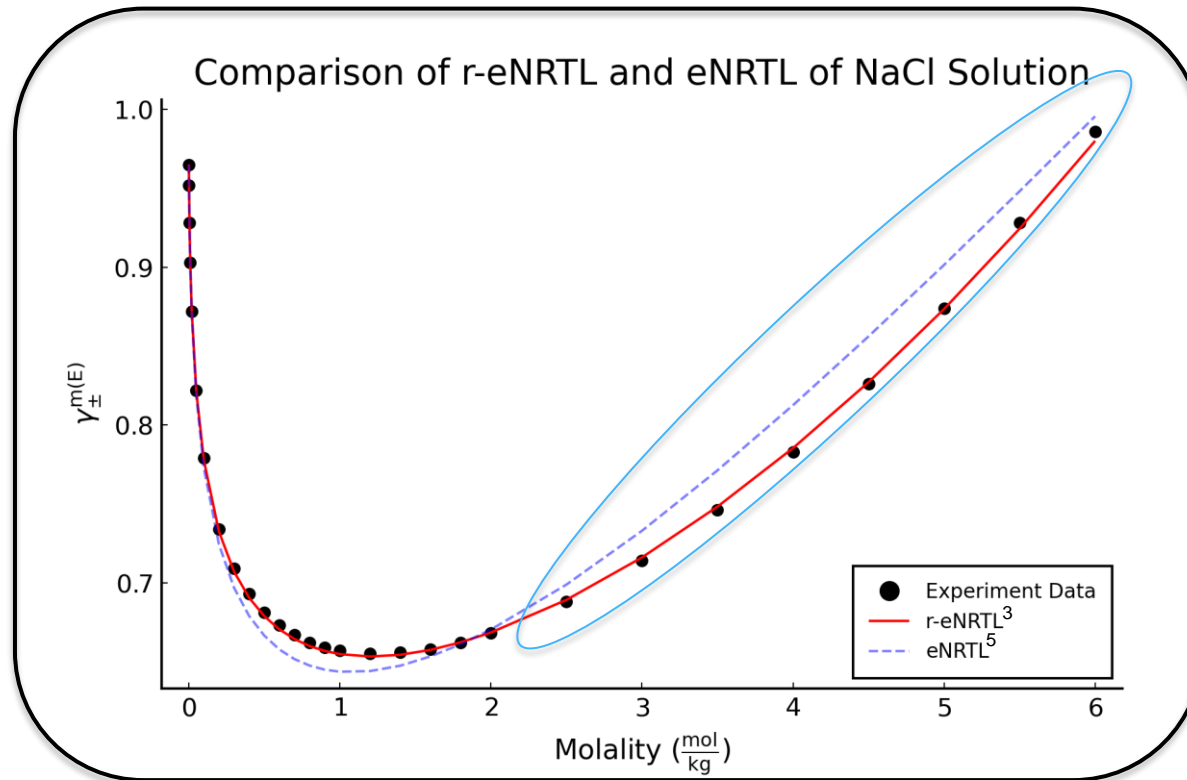
We need accurate mechanistic models that can enable

- Dynamic simulation
- Optimization-based design of water treatment technologies.

[1] Molinari, Raffaele., et al. "Can brine from seawater desalination plants Be a source of critical metals?." *CHEM VIEWS* (2022).

[2] Stuber, Matthew D., et al. "Pilot demonstration of concentrated solar-powered desalination of subsurface agricultural drainage water and other brackish groundwater sources." *Desalination* 355 (2015): 186-196.

Motivation



Modeling accuracy in high concentration regime

➤ Refined e-NRTL³

Modeling system dynamics

➤ ModelingToolkit⁴

[3] Bollas, G.M., et al. Refined electrolyte-NRTL model: Activity coefficient expressions for application to multi-electrolyte systems. *AIChE Journal* 54(6): 1608-1624 (2008).

[4] Ma, Yingbo, et al. Modelingtoolkit: A composable graph transformation system for equation-based modeling. *arXiv preprint arXiv:2103.05244* (2021).

[5] Song, Yuhua, et al. "Symmetric electrolyte nonrandom two-liquid activity coefficient model." *Industrial & Engineering Chemistry Research* 48, no. 16 (2009): 7788-7797.

Refined eNRTL

Short Range Interaction

Long Range Interaction

Born term, for aqueous system = 0

$$G^{*,\text{ex}} = G^{*,\text{SR}} + G^{*,\text{LR}} + \Delta G^{*,\text{Born}}$$
$$= G^{*,\text{SR}} + (A^{*,\text{LR}} + PV) + 0$$

- Increase accuracy of activity coefficient
- Improve accuracy of vapor-liquid equilibrium in high concentration regime for simulating evaporator performance.



Challenges to Implement r-eNRTL

Published paper notation versus computer implementation

- Not in a tensor form
- Hard to get the physical meaning which is important to understanding

Complexity of the model and its use cases

- More complex than original eNRTL⁶ which is already very complex

[6] Chen, Chau-Chyun, et al. "A local composition model for the excess Gibbs energy of aqueous electrolyte systems." *AIChE Journal* 32, no. 3 (1986): 444-454.



New Tensor Notation

Classical Notation⁶

$$\sum_a X_a \sum_c \left(\frac{X_c}{\sum_{c'} X_{c'}} \right) \left(\frac{\sum_j \boxed{X_j} \boxed{G_{ja.ca}} \boxed{\tau_{ja.ca}}}{\sum_j X_j G_{ja.ca}} \right)$$

Tensor Notation

$$\sum_{j=1}^{n_c} X_{c_j} \sum_{k=1}^{n_a} \left(\frac{X_{a_k}}{\sum_{k'=1}^{n_a} X_{a_{k'}}} \right) \left(\frac{\sum_{s \in \{m,a\}} \sum_{l=1}^{n_s} \boxed{X_{s_l}} \boxed{F_{c_j, s_l, a_k}} \boxed{\tau_{c_j, s_l, a_k}}}{\sum_{s \in \{m,a\}} \sum_{l=1}^{n_s} X_{s_l} F_{c_j, s_l, a_k}} \right)$$



Complexity of Refined eNRTL

7

$$\begin{aligned}
 \frac{G^{SR}}{RT} = & \sum_{j=1}^{n_m} X_{m_j} \left(\frac{\sum_{s \in \{m,a,c\}} \sum_{l=1}^{n_s} X_{s_l} F_{m_j, s_l, m_j} \tau_{m_j, s_l, m_j}}{\sum_{s \in \{m,a,c\}} \sum_{l=1}^{n_s} X_{s_l} F_{m_j, s_l, m_j}} \right) \\
 & + \sum_{j=1}^{n_a} X_{a_j} \sum_{k=1}^{n_c} \left(\frac{X_{c_k}}{\sum_{k'=1}^{n_c} X_{c_{k'}}} \right) \left(\frac{\sum_{s \in \{m,c\}} \sum_{l=1}^{n_s} X_{s_l} F_{a_j, s_l, c_k} \tau_{a_j, s_l, c_k}}{\sum_{s \in \{m,c\}} \sum_{l=1}^{n_s} X_{s_l} F_{a_j, s_l, c_k}} \right) \\
 & + \sum_{j=1}^{n_c} X_{c_j} \sum_{k=1}^{n_a} \left(\frac{X_{a_k}}{\sum_{k'=1}^{n_a} X_{a_{k'}}} \right) \left(\frac{\sum_{s \in \{m,a\}} \sum_{l=1}^{n_s} X_{s_l} F_{c_j, s_l, a_k} \tau_{c_j, s_l, a_k}}{\sum_{s \in \{m,a\}} \sum_{l=1}^{n_s} X_{s_l} F_{c_j, s_l, a_k}} \right)
 \end{aligned}$$

$$\log \gamma_{t_j}^{SR} = \frac{\partial}{\partial N_{t_j}} \left(\sum_{\hat{t} \in \{m,a,c\}} \sum_{\hat{j}=1}^{n_{\hat{t}}} N_{\hat{t}_{\hat{j}}} \left(\frac{G^{SR}}{RT} \right) \right)$$

[7] Gottlieb, Robert X., et al. "Automatic Source Code Generation of Complicated Models For Deterministic Global Optimization With Parallel Architectures."

Complexity of Refined eNRTL

$$\frac{G^{\text{SR}}}{RT} = \sum_{j=1}^{n_m} X_{m_j} \left(\frac{\sum_{s \in \{m,a,c\}} \sum_{l=1}^{n_s} X_{s_l} F_{m_j, s_l, m_j} \tau_{m_j, s_l, m_j}}{\sum_{s \in \{m,a,c\}} \sum_{l=1}^{n_s} X_{s_l} F_{m_j, s_l, m_j}} \right)$$

$$+ \sum_{j=1}^{n_a} X_{a_j} \sum_{k=1}^{n_c} \left(\frac{X_{c_k}}{\sum_{k'=1}^{n_c} X_{c_{k'}}} \right) \left(\frac{\sum_{s \in \{m,c\}} \sum_{l=1}^{n_s} X_{s_l} F_{a_j, s_l, c_k} \tau_{a_j, s_l, c_k}}{\sum_{s \in \{m,c\}} \sum_{l=1}^{n_s} X_{s_l} F_{a_j, s_l, c_k}} \right)$$

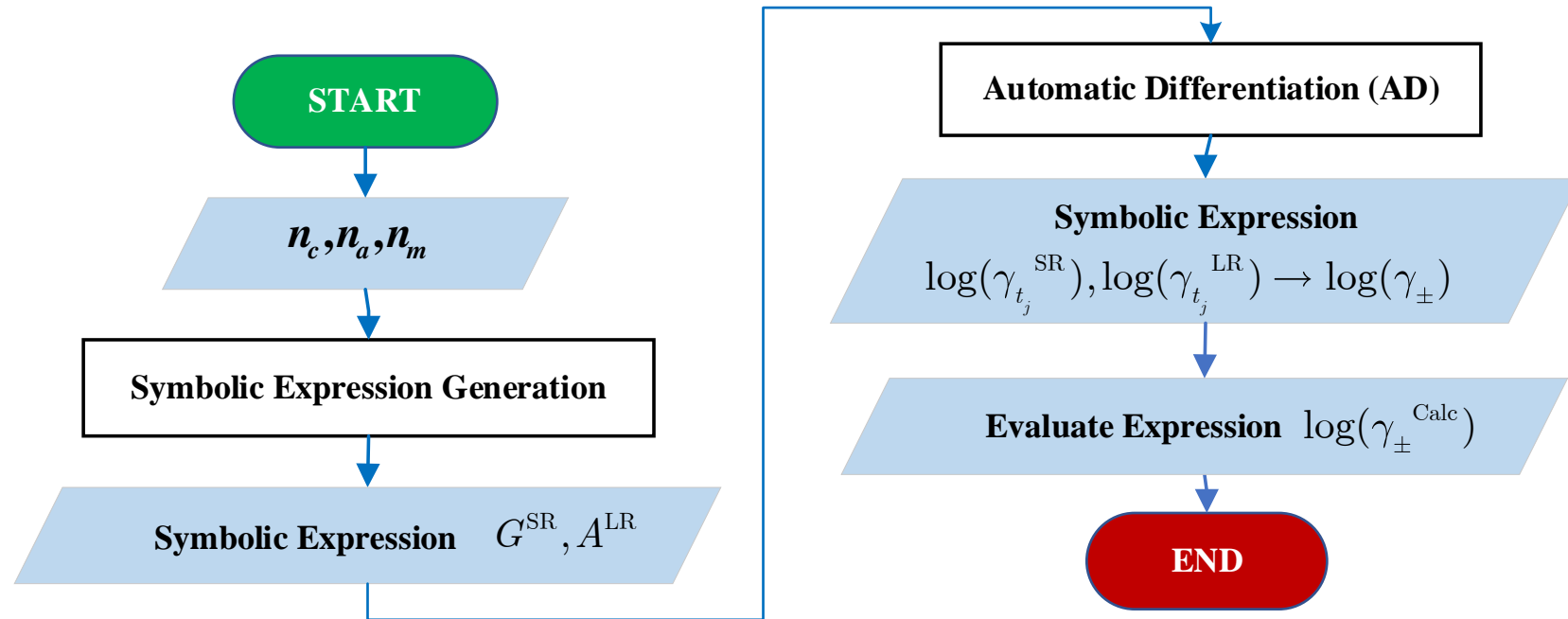
$$+ \sum_{j=1}^{n_c} X_{c_j} \sum_{k=1}^{n_a} \left(\frac{X_{a_k}}{\sum_{k'=1}^{n_a} X_{a_{k'}}} \right) \left(\frac{\sum_{s \in \{m,a\}} \sum_{l=1}^{n_s} X_{s_l} F_{c_j, s_l, a_k} \tau_{c_j, s_l, a_k}}{\sum_{s \in \{m,a\}} \sum_{l=1}^{n_s} X_{s_l} F_{c_j, s_l, a_k}} \right)$$

$$\log \gamma_{t_j}^{\text{SR}} = \frac{\partial}{\partial N_{t_j}} \left(\sum_{\hat{t} \in \{m,a,c\}} \sum_{\hat{j}=1}^{n_{\hat{t}}} N_{\hat{t}_j} \left(\frac{G^{\text{SR}}}{RT} \right) \right)$$

AD

Complicated Expression

Automatic Differentiation Workflow

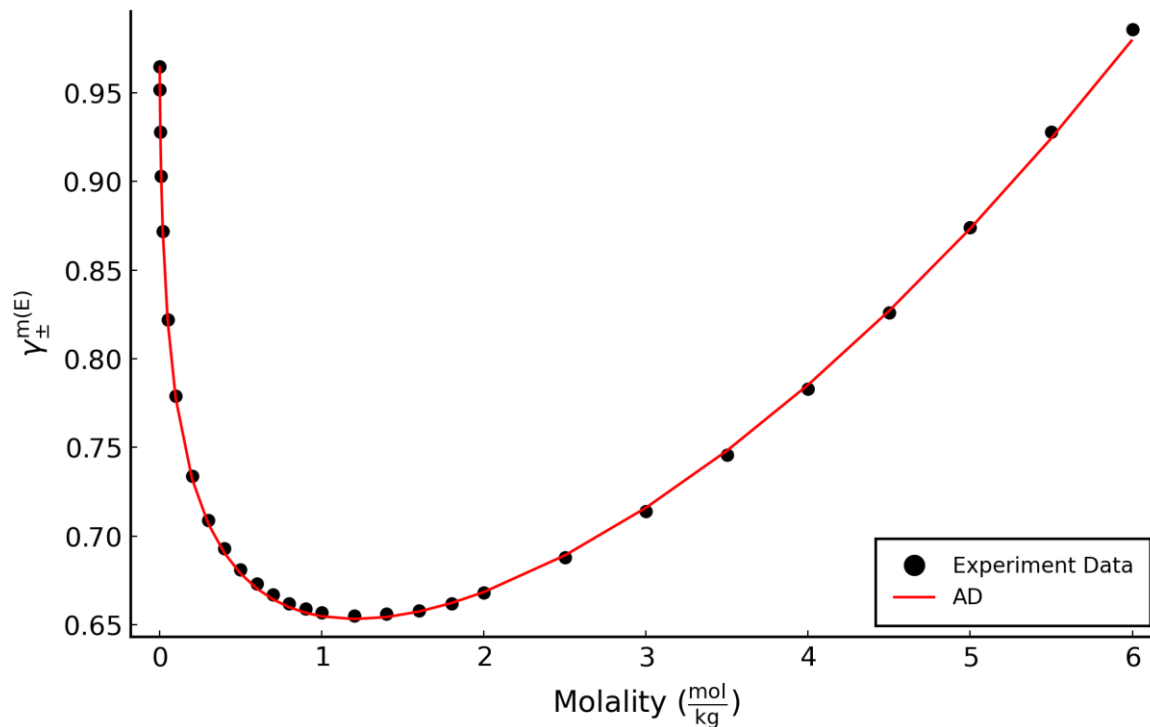


n_c, n_a, n_m : Number of species in aqueous phase.
 G^{SR} : Short range excess Gibbs free energy.
 A^{LR} : Long range excess Helmholtz free energy.

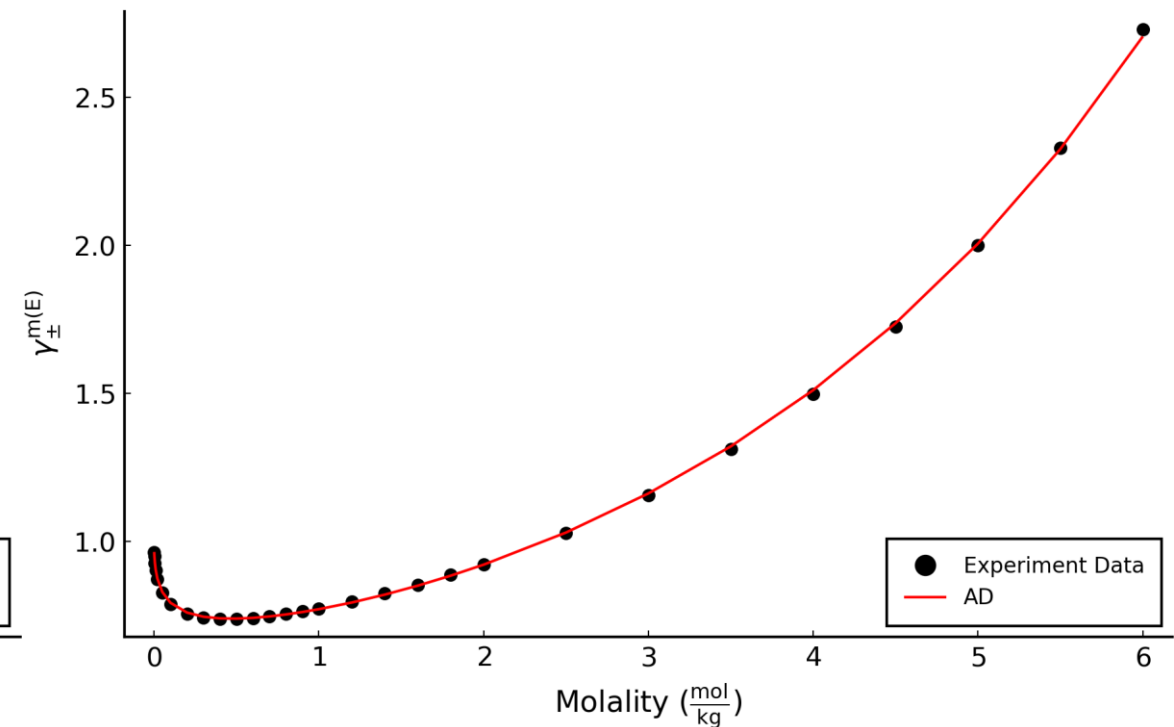
$\gamma_{t_j}^{\text{SR}}$: Short Range activity coefficient of species t_j .
 $\gamma_{t_j}^{\text{LR}}$: Long Range activity coefficient of species t_j .
 t_j : Species with type t , $t \in \{a, c, m\}$ and index j .

Simulation Result with AD

Verification of r-eNRTL calculation NaCl



Verification of r-eNRTL calculation LiCl



Two Systems

ModelingToolkit⁴ (Julia)

Dynamic system modeling using
ModelingToolkit.jl library



Open Source Software for Scientific Machine Learning



<https://sciml.ai>

IDAES (Pyomo)

Steady-state IDAES model under
nonideal equilibrium.



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[9] Stuber, M.D. (2023). NAWIConcentratedElectrolytes. GitHub. https://github.com/PSORLab/NAWIConcentratedElectrolytes/tree/main/flowsheets/old_Modelica



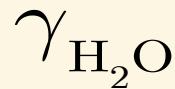
Two Systems

ModelingToolkit⁴ (Julia)

Dynamic system modeling using
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R-eNRTL



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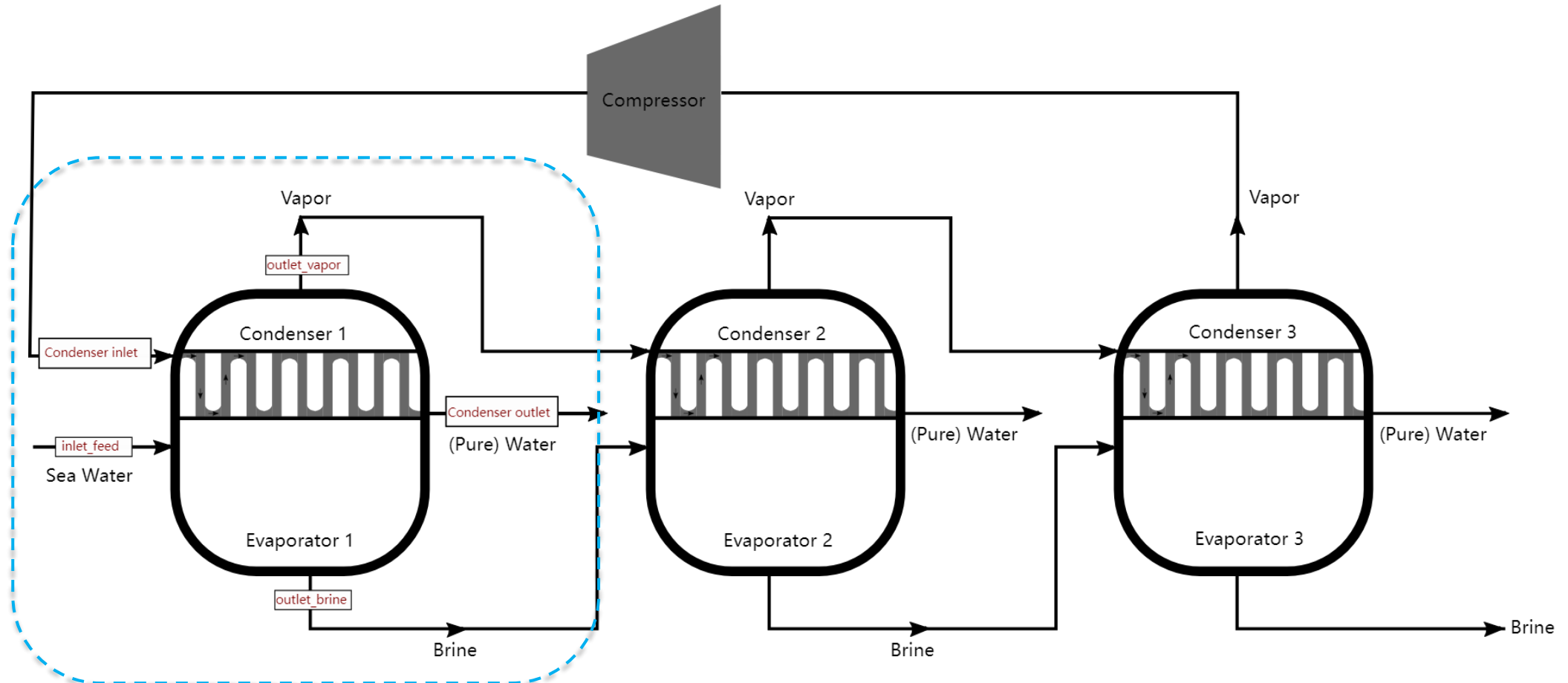


VLE Evaporator

$$x_{\text{H}_2\text{O}} \gamma_{\text{H}_2\text{O}} P_{\text{H}_2\text{O}}^{\text{sat}} = y_{\text{H}_2\text{O}} P_{\text{H}_2\text{O}}$$



Julia Example: 3-Effect MVC-MED

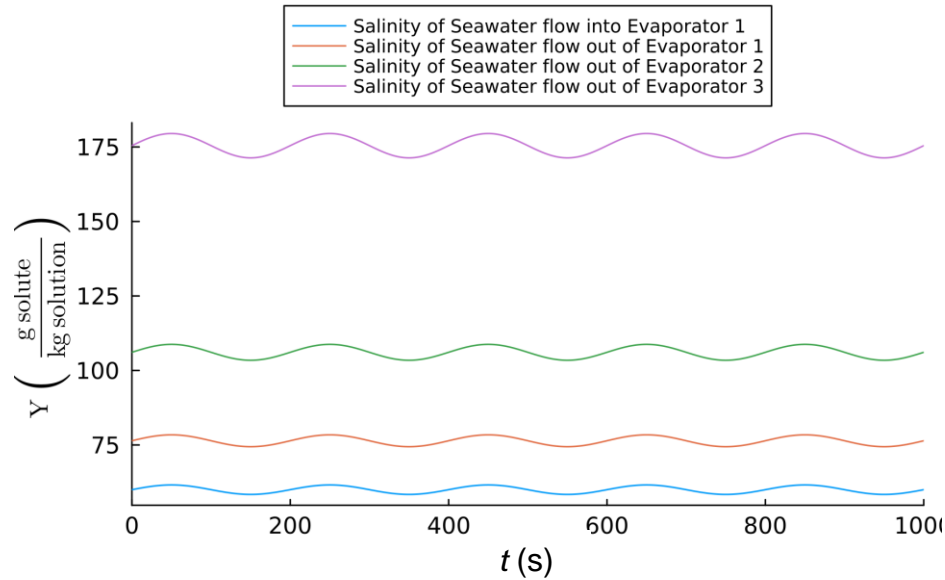


[10] Edna-Soraya, R. (2022). NAWIConcentratedElectrolytes. GitHub. https://github.com/PSORLab/NAWIConcentratedElectrolytes/tree/main/flowsheets/benchmark_system/mvc

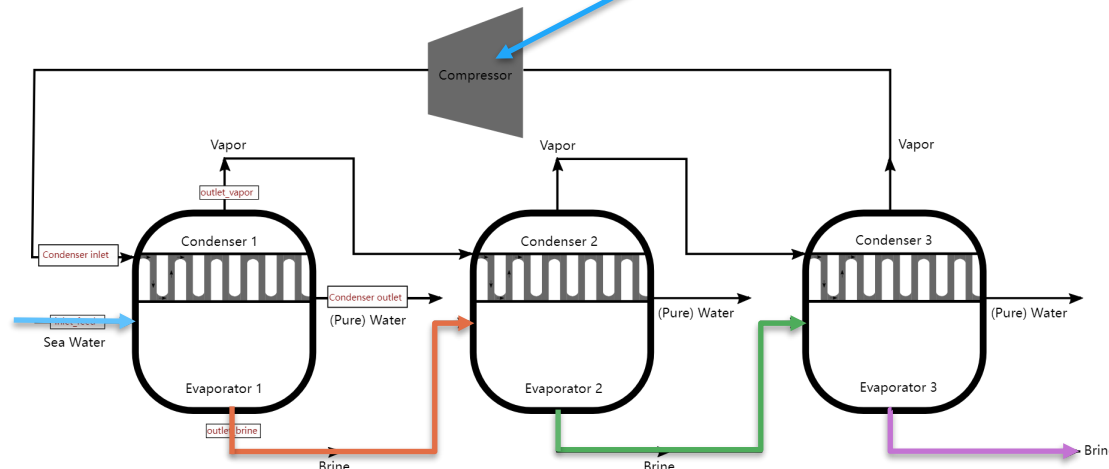
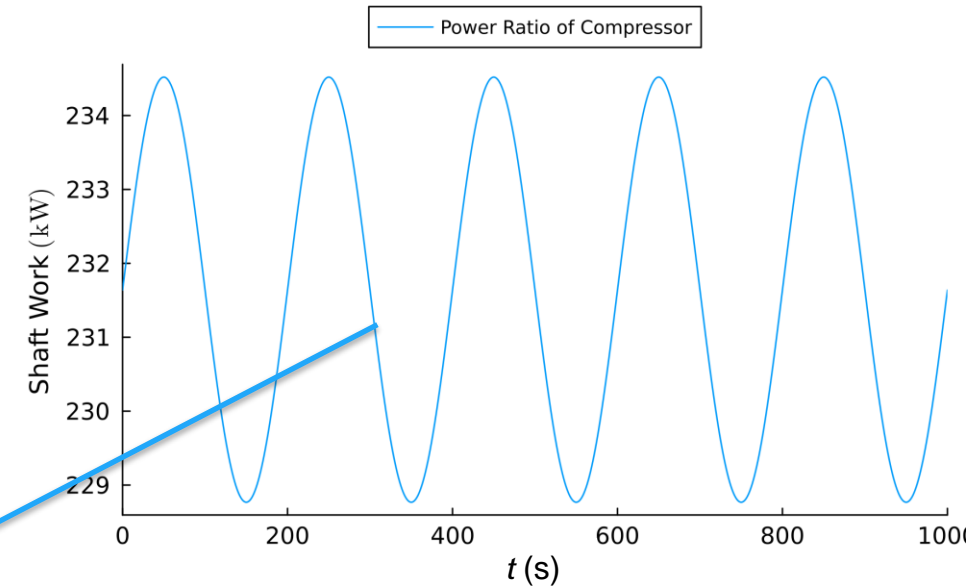


Julia Example: Dynamic Simulation Results

Outlet salinity of 3 evaporators



3 effect MVC shaft work of compressor



IDAES

IDAES (Pyomo)

Steady-state IDAES model under nonideal equilibrium.



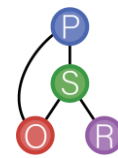
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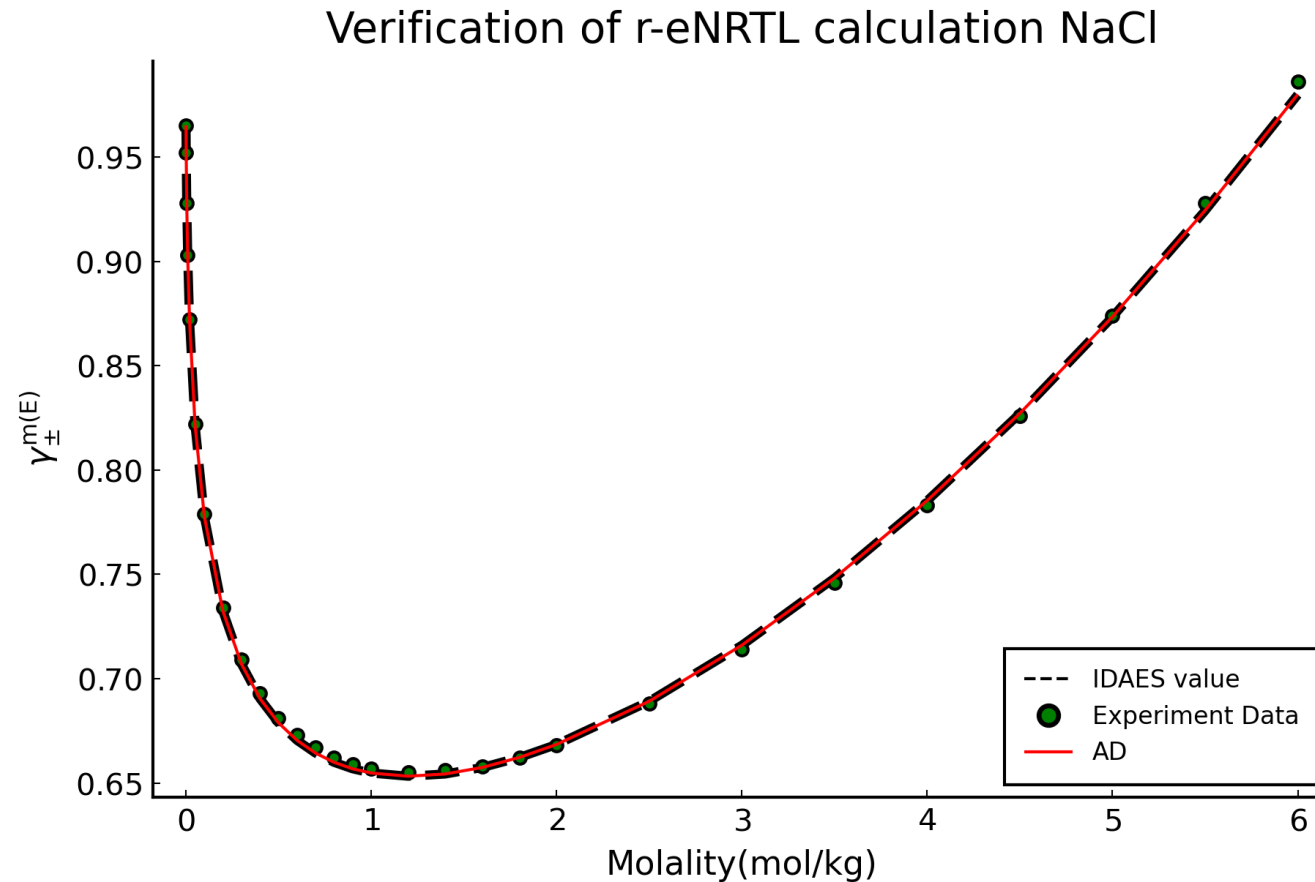
Julia Translated r-eNRTL Model

IDAES 3-effect MVC model

IDAES r-eNRTL model



IDAES vs. Julia r-eNRTL



Transpiler of Symbolic Expression from Julia to Pyomo

Julia

```
[4] ✓ 11.2s Julia  
...  
e  
(1)
```

$$e = \frac{1}{2} \left(1.7745 + \frac{7.951(55.5 - 0.5N_a - 1.5N_c)}{55.5 + 4.4047N_a - 1.5N_c} + \frac{-4.3333 \cdot 10^{19}(N_a + N_c) \left(-0.044516 \left(\frac{0.0010029 + N_a(2.42 \cdot 10^{-5} + \dots)}{\dots} \right) \right)}{\dots} \right)$$

Julia generated the expression and save it into a txt file

Translator

```
line = line.replace('0 + (', '0 * (')  
line = line.replace('g'+ '('', 'g*('')  
expression.append(line)  
  
line_1 = "from pyomo.environ import exp, log"  
line_2 = "def reNRTL_gamma_H2O(Na, Nc):"  
line_3 = "    expr = "+expression[0]  
line_4 = "    return expr"  
  
more_lines = [line_1, line_2, line_3, line_4]  
with open('sympy_script.py', 'w') as f:  
    f.write('\n'.join(more_lines))  
  
from sympy_script import *
```

Translator translate Julia expression into the python function syntax and then import it.

IDAES

```
@m.fs.Constraint(m.fs.set_evaporators, doc="e")  
def eq_nonideal_activity_coefficient(b, e):  
    x = b.evaporator[e].properties_brine[0].m  
    x_m = x*58.443/(x*58.443+(1-x)*18.01528)  
    return (  
        b.act_coeff[e]  
        == reNRTL_gamma_H2O(x_m, x_m)  
    )  
  
# Deactivate equilibrium equation from evapor  
# new equilibrium equation as a Constraint th  
# activity coefficient. Note that since water  
# participating in the vapor-liquid equilibri  
# coefficient and vapor pressure are of water  
# solvent.
```

Call the function inside equality constraint where the activity coefficient are defined

Generated Function in IDAES

```
# Save the calculated activity coefficient in the global
# activity coefficient variable.
@m.fs.Constraint(m.fs.set_evaporators, doc="eNRTL activity coefficient for water")
def eq_nonideal_activity_coefficient(b, e):
    x = b.evaporator[e].properties_brine[0].mole_frac_phase_comp["Liq", "TDS"]
    x_m = x*58.443/(x*58.443+(1-x)*18.01528)
    return (
        b.act_coeff[e]
        == reNRTL_gamma_H2O(x_m,x_m)
    )

# Deactivate equilibrium equation from evaporator and include a
# new equilibrium equation as a Constraint that includes the
# activity coefficient. Note that since water is the only solvent
# participating in the vapor-liquid equilibrium, the activity
# coefficient and vapor pressure are of water as the
# solvent.
```

In the MVC model, we integrate our generated function into the equality constraint, replacing the embedded refined eNRTL model in IDAES.



Comparison between Embedded Function and Translated Function

Julia Translated Function

Other variables:

```
Energy input (kW): 304.4402  
Total water produced (gal/min): 107.0689  
Specific energy consumption (SC, kWh/m3): 12.5191  
Water recovery (%): 70.0000  
Molal conc solute evap 1 (mol/kg): 1.4557  
Molal conc solute evap 2 (mol/kg): 1.9279  
Molal conc solute evap 3 (mol/kg): 3.8498
```

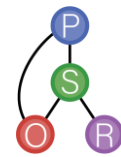
IDAES Embedded Function

Other variables:

```
Energy input (kW): 304.5390  
Total water produced (gal/min): 107.0689  
Specific energy consumption (SC, kWh/m3): 12.5232  
Water recovery (%): 70.0000  
Molal conc solute evap 1 (mol/kg): 1.4555  
Molal conc solute evap 2 (mol/kg): 1.9275  
Molal conc solute evap 3 (mol/kg): 3.8498
```

* Results may be slightly different due to numerical calculations

[11] Edna-Soraya, R. (2022). **watertap-renrtl**. GitHub. <https://github.com/watertap-org/watertap-renrtl>



Conclusion

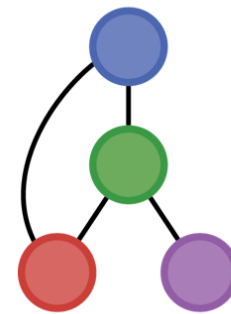
- Implemented refined eNRTL model for single electrolyte with AD.
 - Developed a new notation for r-eNRTL
 - Generate activity coefficients as symbolic expressions
- Built a library to model dynamic evaporator systems that uses r-eNRTL in VLE expressions in Julia.
- Created a toolchain that generates callback functions for IDAES and Pyomo to use the r-eNRTL activity coefficients (and the same procedure can be extended to other properties)
- Currently implementing the multi-electrolyte form and expanding the AD work for all other thermodynamic properties



Thanks!

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Members of the Process Systems and Operations Research Laboratory at the University of Connecticut (<https://psor.uconn.edu/>)



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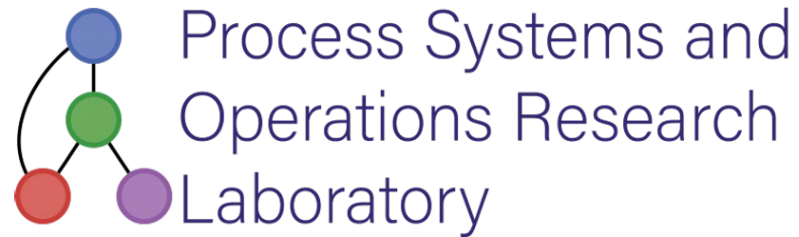
Funding:

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Any opinions, findings, and conclusions or recommendations expressed in this material are those of the authors and do not necessarily reflect the views of the National Alliance for Water Innovation or the U.S. government.



Questions?



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